## **READING GUIDE\***

## **Processes of Soil Formation II: Weathering and Mineral Transformations**

see Buol et al. (2011), Chapter 4, p. 141-161

**OBJECTIVE**: To understand the common weathering reactions that occur in soil environments, recognize the factors that influence these processes, and know how weathering products affect soil properties.

- 1. What is *weathering*? What is the distinction between *geochemical* weathering and *pedochemical* weathering? Why does weathering occur?
- 2. What are primary minerals? What are secondary minerals?
- 3. What is *physical weathering*? What processes are associated with physical weathering? What is *chemical weathering*? What processes are associated with chemical weathering?
- 4. What is *oxidation*? Describe an example of chemical weathering by oxidation that occurs in soils. What is *reduction*? What is *gleization*? Describe an example of chemical weathering by reduction that occurs in soils.
- 5. Which soil constituents are susceptible to oxidation-reduction reactions? Which are not?
- 6. What soil morphologies are associated with alternating oxidizing and reducing conditions? How do these morphologies differ if the pore is the site of reduction or the site of oxidation?
- 7. How does pH influence the stability of Fe minerals?
- 8. What is *ferrolysis*? Describe an example of chemical weathering by ferrolysis that occurs in soils.
- 9. What is *hydration*? Describe an example of chemical weathering by hydration that occurs in soils.
- 10. What is *hydrolysis*? Describe an example of chemical weathering by hydrolysis that occurs in soils.
- 11. What are the products of the hydrolysis of primary minerals in soils? What is the significance of each of these products within the soil system?
- 12. What is solution? Describe an example of chemical weathering by solution that occurs in soils.
- 13. What is *chelation*? Describe an example of chemical weathering by chelation that occurs in soils.
- 14. Describe the processes by which mica is weathered through removal of interlayer potassium. What conditions promote this process?
- 15. What is a *hydroxy interlayered mineral* (HIM)? How do HIM form in soils? How can HIM formation influence other soil properties? What is gibbsite? What conditions in the soil favor gibbsite formation?
- 16. For ferric and ferrous forms of iron, which is oxidized and which is reduced? What is the ionic charge of each?
- 17. Which iron minerals are precipitated under oxidizing conditions when Fe is released from primary minerals? What iron mineral can form in the soil through transformation of the other iron minerals?
- 18. Among primary minerals, how does the order in which they crystallize from molten rock compare to the order in which they weather in soil environments?
- 19. What properties of primary (sand- and silt-sized) minerals influence their stability in soil environments? (Note: There are three.) What properties of secondary (clay-sized) minerals influence their stability in soils?
- 20. For each of the secondary (clay-sized) minerals, describe the conditions that promote their presence, formation, or stability in soil environments.

<sup>\*</sup> Questions in plain type represent basic facts and concepts. Questions in **bold** type are those that are answered in the text but require more careful consideration. The Synthesis questions at the end help you apply the facts and concepts to a relevant issue.